

Facile construction of spirobicyclic skeletons by intramolecular aldol reaction: simple formal syntheses of (±)-spirojatamol and (±)-erythrodiene

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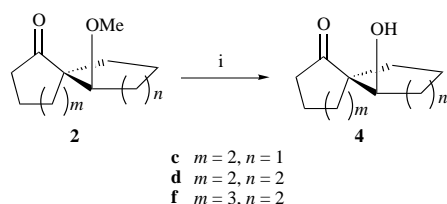
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An intramolecular aldol reaction of keto acetals **1**, regioselectively performed with Me₃SiI–(Me₃Si)₂NH leading to spirobicyclic compounds **2** and **3**, has been applied to the syntheses of (±)-spirojatamol and (±)-erythrodiene.

The aldol reaction is one of the most versatile methods for carbon–carbon bond formation in organic synthesis.¹ In particular, the intramolecular version is a useful method for the synthesis of a variety of ring structures. A new methodology for the construction of polycyclic ring systems fused to cyclobutane has recently been developed using a tandem intramolecular Michael–aldol reaction, carried out with Me₃SiI in the presence of (Me₃Si)₂NH.² Here we report a regioselective intramolecular aldol reaction of keto acetals **1** using Me₃SiI–(Me₃Si)₂NH and the successful use of this methodology for syntheses of (±)-spirojatamol and (±)-erythrodiene.

The results of the intramolecular aldol reaction of keto acetals **1**† are summarized in Table 1. Reaction of **1a** with Me₃SiI–(Me₃Si)₂NH in CH₂Cl₂ at 0 °C gave a mixture of spiro compounds **2a** and **3a** in 78% yield. Thus, the aldol reaction regioselectively takes place at the more substituted α-carbon atom. Using the same procedure, **1b–f** were converted into the corresponding spiro compounds **2b–f** and **3b–f** in good yields.

The stereostructures of **2c**, **d** and **f** were determined by their transformation into the known alcohols **4c**, **d** and **f**‡ by treatment with BBr₃ (Scheme 1)‡ while the structures of the other products were assigned by spectral analysis.



Scheme 1 Reagents and conditions: i, BBr₃, CH₂Cl₂, –78 to –40 °C

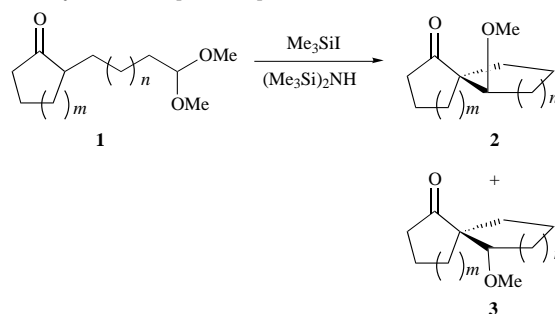
Spirojatamol **5**, isolated from the roots of *Naldostuchys jata-mansi*,⁴ and erythrodiene **6**, found in the encrusting Caribbean gorgonian coral *Erythropodium caribaeorum*,⁵ have a spiro-[4.5]decane framework. Recently, Forsyth^{6a} and we^{6b} independently accomplished total syntheses of these natural products. The present approach for the preparation of these natural products has been designed by way of the above intramolecular aldol reaction.

N-Cyclohexyl-4-isopropylcyclohexanimine **7**^{6b} was treated with lithium diisopropylamide (LDA) in the presence of hexamethylphosphoric triamide (HMPA) and then with 4-bromo-

† Keto acetals **1a–f** were prepared from the corresponding cyclic ketones in two steps; formation of the cyclohexyl imines, followed by alkylation.

‡ The isomerization at the C-1 and C-5 positions of **4c** occurred during the conversion.

Table 1 Synthesis of spiro compounds **2** and **3** from keto acetals **1**



Substrate	Yield (%)	Ratio 2 : 3
1a (<i>m</i> = 1, <i>n</i> = 1)	78	1.1 ^a
1b (<i>m</i> = 1, <i>n</i> = 2)	70	1.2 ^b
1c (<i>m</i> = 2, <i>n</i> = 1)	83	1.6 ^a
1d (<i>m</i> = 2, <i>n</i> = 2)	92	2.4 ^a
1e (<i>m</i> = 3, <i>n</i> = 1)	87	1.5 ^a
1f (<i>m</i> = 3, <i>n</i> = 2)	89	1.5 ^a

^a Isolated ratio. ^b Determined by integration of peaks in the ¹H NMR spectrum.

butanal dimethyl acetal to give ketone **8** as a mixture of two diastereoisomers (Scheme 2). The aldol reaction of **8** with Me₃SiI–(Me₃Si)₂NH furnished, in excellent yield, a mixture of **9a–d** in the ratio of 2.4:1.3:1.0:1.2. After separation of the four diastereoisomers, **9a–d** were demethylated with BBr₃. Though isomerization at the C-1 and C-5 positions of the products occurred to some extent during the transformation, alcohols **10a–d**§ were obtained as the major products from the corresponding stereoisomers, respectively. Treatment of **10a** and **b** with tetrapropylammonium perruthenate (TPAP) and *N*-methylmorpholine *N*-oxide (NMO)⁷ provided **11** which has been correlated with (±)-spirojatamol **5** and (±)-erythrodiene **6**.^{6b} On the other hand, oxidation of **10c** and **d** gave diketone **12**.

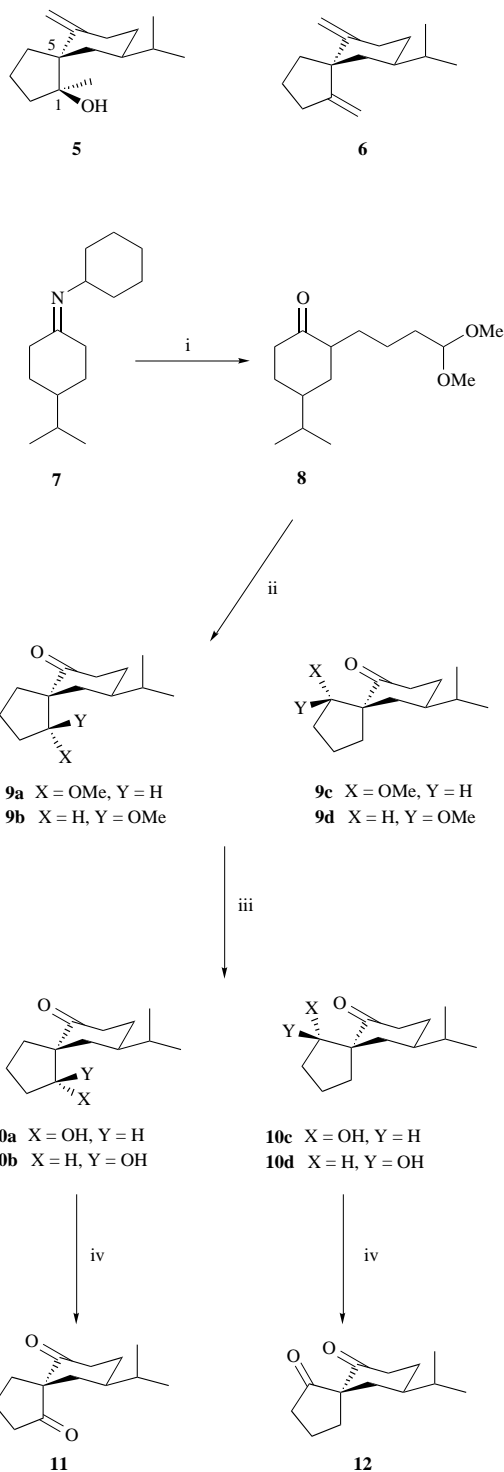
The above method conducted under mild reaction conditions should be useful for the assembly of spiro skeletons and for syntheses of various natural products.

Experimental

Typical reaction procedure for 1-methoxyspiro[4.5]decan-6-ones **2c** and **3c**

To a stirred solution of the keto acetal **1c** (0.837 g, 3.91 mmol) and (Me₃Si)₂NH (1.24 cm³, 5.87 mmol) in dry CH₂Cl₂ (25 cm³) at 0 °C was added Me₃SiI (0.67 cm³, 4.70 mmol) and the mixture was stirred for 30 min at the same temperature. After addition of 10% hydrochloric acid followed by stirring of the mix-

§ The stereostructures of **9a** and **9b** were assigned by comparison with spectral data for **2c** and **3c**. On the other hand, that of **10d** was established on the basis of the intramolecular hydrogen bond between the hydroxy group and the carbonyl group.



Scheme 2 Reagents and conditions: i, LDA, HMPA, 4-bromobutanol dimethyl acetal, THF, 0 °C, 88%; ii, Me₃SiI, (Me₃Si)₂NH, CH₂Cl₂, 0 °C, 94%; iii, BBr₃, CH₂Cl₂, -78 to -40 °C, 50–61%; iv, TPAP, NMO, 4 Å molecular sieves, CH₂Cl₂, room temp., 85–96%

ture for 30 min, the resulting mixture was extracted with diethyl ether. The extract was washed with 10% aqueous sodium thiosulfate and brine, dried over potassium carbonate and the solvent evaporated. The residue was subjected to chromatography on silica gel with ethyl acetate–hexane (1:6 v/v) to give **3c** (0.228 g, 32%) as an oil and **2c** (0.366 g, 51%) as an oil.

3c: ν_{\max} (neat)/cm⁻¹ 1720; δ_{H} (300 MHz; CDCl₃) 1.42–1.95 (12 H, m), 2.35–2.42 (2 H, m, 7-H₂), 3.30 (3 H, s, OMe), 4.07 (1 H, br t, *J* 7, \ddagger 1-H); δ_{C} (75 MHz; CDCl₃) 20.6, 22.0, 27.0, 29.4, 32.0, 34.1, 39.8, 57.5, 59.7, 83.5, 214.0; *m/z* 182 (M⁺) (HRMS: found M⁺, 182.1299. C₁₁H₁₈O₂ requires *M*, 182.1307).

2c: ν_{\max} (neat)/cm⁻¹ 1706; δ_{H} (300 MHz; CDCl₃) 1.04–1.14 (1 H, m), 1.38–1.90 (10 H, m), 2.01–2.10 (1 H, m), 2.35–2.61 (2 H, m, 7-H₂), 3.87 (3 H, s, OMe), 3.84–3.89 (1 H, m, 1-H); δ_{C} (75 MHz; CDCl₃) 21.2, 22.6, 27.7, 29.1, 32.0, 38.4, 41.6, 56.6, 61.3, 87.1, 211.5; *m/z* 182 (M⁺) (HRMS: found M⁺, 182.1308).

Acknowledgements

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\ddagger *J* Values given in Hz.

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